

Spaces Between Particles

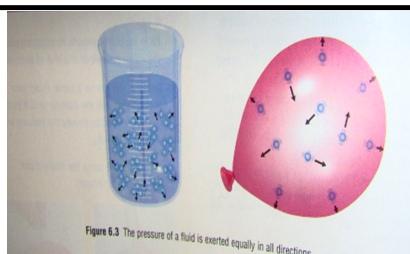
Space between particles depends on 2 factors:

- 1) the physical state of the substance (solid, liquid, or gas)
- 2) the amount of energy that the particles have

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Since gas particles are spread out they have lots of space between particles, thus gases are compressible.

Compressible - is the ability to be squeezed into a smaller volume



When force is applied to a solid or liquid the particles within cannot move closer so they transmitted (pass along) the force like the domino effect. They are incompressible.



Gases are easiest to compress, solids most difficult | Compressibility | Chemistry

When gases are under pressure (squeezed in a bottle), they are ready to expand when given the opportunity. So when a gas finds a way to escape, through a hole or nozzle, it exits with a great deal of force.

Great for oxygen tanks and certain tools such as air compressors

Flow Pressure - is pressure that causes motion because the fluid is moving

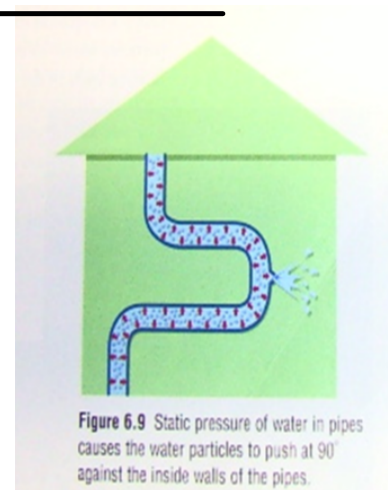
ex) wind is moving air that can lift your hair

ex) water pressure from tap can remove food off plates

Static Pressure - is a fluid's pressure that exerts a force on an object even if it is not moving. It pushes 90° to the walls of the pipe.

ex) When the water in your tap is turned off, the water in your pipes is exerting static pressure.

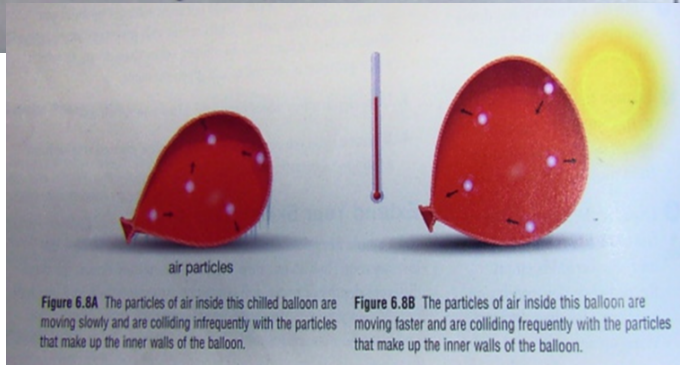
ex) when you swim deep underwater you can feel pressure on your eardrums. (the deeper you go the more pressure you feel since the weight of the water and air above you pushes down.)



Pressure and Temperature

The particle theory suggests that particles move faster when they are heated because they gain more energy. Keeping the particle theory in mind, compare Figures 6.8A and B, representing the behaviour of air particles inside two balloons.

- In which balloons are the particles of air colliding with each other and with the inner walls of the balloon with greater force?
- Pressure is force measured over a certain area. In which balloon are the particles of air exerting greater pressure against the inner walls of the balloon?
- Does an increase in temperature cause an increase or a decrease in pressure?



Increase heat causes an increase in pressure

Pressure in fluids is important in hydraulic machinery

Hydraulics - is the study of pressure in liquids

Hydraulic Systems - are devices that transmit applied forces through a liquid to move something else.

